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Student Name: Pd. Date: Stoichiometry Word Problems 1 - SOLUTIONS 1. The emergency airbags in modern automobiles use a rapid chemical reaction to fill the bag in the case of an accident. Sodium azide (Na₃N) reacts with potassium nitrate to rapidly produce nitrogen gas, potassium oxide, and sodium oxide. The balanced (you're welcome) reaction is as follows: $10 \text{Na}_3\text{N} + 2 \text{KNO}_3 \rightarrow 16 \text{N}_2 + 5 \text{Na}_2\text{O} + 2 \text{K}_2\text{O}$ The ...





















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