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AP Chemistry Chapter 5 Answers Zumdahl 5.27a.  $4.8 \text{ atm} \times 760 \text{ mmHg} / 1 \text{ atm} = 3.6 \times 10^3 \text{ mmHg}$ .  $3.6 \times 10^3 \text{ mmHg} \times 1 \text{ torr} / 1 \text{ mmHg} = 3.6 \times 10^3 \text{ torr}$ .  $4.8 \text{ atm} \times 1.013 \times 10^5 \text{ Pa} / 1 \text{ atm} = 4.9 \times 10^5 \text{ Pa}$ .  $4.8 \text{ atm} \times 14.7 \text{ psi} / 1 \text{ atm} = 71 \text{ psi}$ .  $296.5 \text{ cm} \times 10 \text{ mm} / 1 \text{ cm} = 65 \text{ mmHg} = 65 \text{ torr}$ ;  $65 \text{ torr} \times 1 \text{ atm} / 760 \text{ torr} = 8.6 \times 10^{-2} \text{ atm}$ .  $6.6 \times 10^{-2} \text{ atm} \times 1.013 \times 10^5 \text{ Pa} / 1 \text{ atm} = 8.7 \times 10^4 \text{ Pa}$ . 5.31 If the levels of Hg in each arm of the m...

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AP Chemistry--Chapter 13: Chemical Equilibrium Practice Problems 1. Write the equilibrium expression for the following reaction:  $4\text{NH}_3(\text{g}) + 7\text{O}_2(\text{g}) \rightleftharpoons 4\text{NO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{g})$  2. The following equilibrium concentrations were observed for this reaction at 127C:  $[\text{N}_2(\text{g})] = 3.1 \times 10^{-2} \text{ mol/L}$ ,  $[\text{NH}_3] = 8.5 \times 10^{-1} \text{ mol/L}$ ,  $[\text{H}_2] = 3.1 \times 10^{-3} \text{ mol/L}$  a. Calculate the value of K at 127C for this reaction. ...

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For the following problems assume that there is no air resistance. When solving the 2 nd or b part of a problem, DO NOT use the answer from the 1st or a part. 1) Studies have shown that a fall from 5 ft can result in a broken bone. a) From this height, how long will it take you to reach the ground? b) From this height, what will your velocity be just before you hit the ground? 2) A rock is dropped ...

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Note: Highlighted items are the concepts addressed on the AP Exam, other information given is to compliment the AP Chemistry--Chapter 21: Nuclear Chemistry subject and outline information presented in the text in Chapter 21. Example Problems 1. What product is formed when radium-226 undergoes alpha decay? What isotope undergoes alpha decay to form lead-208? 2. Write nuclear equations for the followin...

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